## **B.Sc. BIOTECHNOLOGY** FIRST SEMESTER [SPECIAL REPEAT] BIOCHEMISTRY & METABOLISM

**BBT-101** 

[USE OMR SHEET FOR OBJECTIVE PART]

d. Both hydrophilic and hydrophobic

Duration: 3 hrs.

**Objective** 

Time: 30 mins.

Marks: 20

Full Marks: 70

SET

Choose the correct answer from the following:

 $1 \times 20 = 20$ 

- 1. The backbone of DNA is:
  - a. Hydrophilic
  - c. Neutral
- Molecular formula of pamitic acid is:
- a. C<sub>10</sub>H<sub>20</sub>O<sub>2</sub>
  - c. C<sub>12</sub>H<sub>24</sub>O<sub>2</sub>

b. C<sub>16</sub>H<sub>32</sub>O<sub>2</sub>

b. Hydrophobic

- d. C<sub>8</sub>H<sub>16</sub>O<sub>2</sub>
- What is produced in the light reaction of photosynthesis?
  - a. ATP and NADPH b. Glucose
  - c. CO2 d. H<sub>2</sub>O
- 4. Ramachandran plot is used for:
  - a. Predicting the structure of an enzyme
  - c. Predicting the secondary of proteins from primary sequence
- b. Predicting the structure of a protein d. All the above

b. Enzyme kinetics

- 5. LB plot is used to determine:
  - a. Specificity of an enzyme
  - d. Inborn errors of metabolism c. Nomenclature of an enzyme
- 6. The basic repeating unit of a DNA molecule is:
  - a. Nucleoside
    - b. Nucleotide c. Histones d. Amino acids
- During one Kreb cycle number of carbondioxide molecules released is:
  - a. 1

b. 2

- 8. Cyclic photophosphorylation results in the formation of:
  - a. ATP

- b. NADPH
- c. ATP and NADPH
  - Lyases,a class of enzyme, catalyses:
    - a. Hydrolysis reaction c. Oxidation and reduction reaction
- b. Group transfer reaction

d. ATP, NADPH and Oxygen

d. Addition of groups to double bond and vice versa

10.	The overall equation for the aerobic cellular respiration of glucose is:  The overall equation for the aerobic cellular respiration of glucose is:  The overall equation for the aerobic cellular respiration of glucose is:			
	a. $CO_2+H_2O \rightarrow C_6H_{12}O_6+O_2+ATP+Heat$	0. 6111206.02		
	c. C <sub>6</sub> H <sub>12</sub> O <sub>6</sub> →Lactic acid+ATP+Heat	d. $C_6H_{12}O_6 \rightarrow CO_2 + Ethyl$		
		alchohol+ATP+Heat		
11	What is the nature of an enzyme?			
11.	a. Vitamin	b. Lipid		
	c. Carbohydrate	d. Protein		
	. Out of the following the one having highest redox potential is:			
12.		b. O <sub>2</sub>		
	a. Ubiquinone c. FMN	d. NAD		
13.	All of the reactant will be converted to pro			
	a. Will never reach equilibrium	b. Will not occur spontaneously		
	c. Will proceed at a rapid rate	d. Will proceed at a rapid rate		
14.	14. What is the maximum wavelength that Tryptophan and tyrosine absorb?			
	a. 280nm	b. 260nm		
	c. 257nm	d. 230nm		
15	Sphingomyelins are found in:			
10.	a. Muscles	b. Nephrons		
	c. Brain tissues	d. Hepatocytes		
16	Metal ions that temporary binds substrate			
	a. Inhibitors	b. Coenzymes d. Cofactors		
	c. Prosthetic group	•		
17.	<ol><li>Glycerol is required for the formation of all the following except:</li></ol>			
	a. Glucose	b. Triacylglycerol		
	c. Phospholipids	d. Glycolipids		
18.	If energy releases excessively in environment	ent, having less energy products than		
	reactants, resulting reaction is called:			
	a. Redox reaction	b. Thermodynamics		
	c. Exergonic reaction	d. Endergonic reaction		
19	Arrangement of nucleotides in DNA can be seen by:			
15.	a. Ultracentrifuge	b. X-Ray crystallography		
	c. Light microscope	d. Electron microscope		
20.	A short length of DNA molecule has 80 thiamine and 80 guanine bases. The total			
	number of nucleotide in the DNA fragmen			
	a. 160	b. 40		
	c. 320	d. 640		

## (<u>Descriptive</u>)

Time: 2 hr. 30 mins.		Marks: 50
	[ Answer question no.1 & any four (4) from the rest ]	
1.	Define carbohydrates. Write a note on their classification.	3+7=10
2.	What are enzymes? Describe their classification and nomenclature.	3+7=10
3.	<ul><li>a) Write short note on zwitter ions.</li><li>b) Write short note on peptide bonds.</li></ul>	5+5=10
4.	Describe in detail: a) Embden Mayorhoff Pathway (EMP) b) TCA cycle	5+5=10
5.	Define proteins. Write a note on their classification.	6+4=10
6.	<ul><li>a) What are lipids? How are they classified?</li><li>b) Write the reaction involved when fatty acid is reacted with alkali.</li></ul>	5+5=10
7.	<ul><li>a) Write the structure of a DNA and write a note on the two type bonds present within them to maintain their structure.</li><li>b) What are the various factors responsible for the denaturation of DNA?</li></ul>	10
8.	Write a short note on: a) Vitamin B12 b) Lipoic acid	5+5=10

== \*\*\* = =