REV-00 MPH/68/80

# M. Sc. Physics FIRST SEMESTER QUANTUM MECHANICS -I MPH - 103

Duration: 3 Hrs.

Marks: 70

Part : A (Objective) = 20 Part : B (Descriptive) = 50

## [<u>PART-B: Descriptive</u>]

Duration: 2 Hrs. 40 Mins.

### Marks: 50

#### [Answer question no. One (1) & any four (4) from the rest ]

- 1.a) Explain with proper examples the meaning of wave particle<br/>duality. Discuss how Schrodinger established the validity of it.3+2=5
  - b) (i) Show that the wavelength of the quantum wave associated with the electron accelerated through a potential difference of 150 volts lies in X-ray range.

(ii)Calculate momentum of the electron of 100 electron volt 2+3=5 energy.

- Write down the Schrodinger wave equation for a particle of mass 'm' 2+3+2+3 confined in a box of length 'a' such that V=0 for 0 ≤ x ≤ a and v=α for =10 x ≤ 0 and ≥ a. Deduce the expression for wave function and energy. Calculate ground state energy of the electron in a box of size 10<sup>-14</sup> meter in S.I unit.
- a) What do you mean by Hermitian operator? Prove that two 1+4=5 eigen function of a Hermitian operator belonging to different eigen values are orthogonal.
  - b) (i) Examine if d<sup>2</sup>/dx<sup>2</sup> is a Hermitian operator. 2+3=5
     (ii)Calculate the average value of the momentum of a particle confined in a length 'a'.

2017/12

What do you mean by identical particles? Distinguish between classical 2+2+2+1 and quantum identical particles. Define particle exchange operator and +3=10 calculate the eigen values of the particle exchange operator.

3+2=5

- 5. (a) Evaluate the following commutators.
  - I.  $[\hat{L}_+, \hat{L}_-]$ II.  $[\hat{L}_Z, \hat{L}_\pm]$

(b) Find the equation of motion in Heisenberg representation.

- (a) Show that spherical harmonics are eigen functions of L<sup>2</sup> and L<sub>z</sub>. 3+2=5
   Write the eigen values of both the operators.
  - (b) (i) Give the relation between the spin angular momentum and spin magnetic moment. What are pauli's spin matrices?

(ii) Prove that  $[\sigma_x, \sigma_y] = 2i\sigma_z$ 

7. (i) What do you mean by symmetric and antisymmetric wave function? 1+1+2+1
 Explain exchange energy. Discuss how spin statistics are connected with +1=6
 symmetric and antisymmetric wave functions.

(ii) How can you contruct symmetric and antisymmetric wave function 2+2=4 from unsymmetric wavefunction ?

 (i) Define radial probability density of finding the electron of the hydrogen atom at a distance 'r' and write the expression of it. Calculate the position of maximum probability of finding the electron in its ground state.

(ii) A proton is confined to a nucleus of radius  $5 \times 10^{-15}$  m. Calculate 2+2=4 the uncertainty in its momentum and minimum kinetic energy of the proton (mass of the proton is  $1.67 \times 10^{-27}$  kg).

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## [ PART-A : Objective ]

### Choose the correct answer from the following:

1×20=20

- 1. The acceptable wave function is
  - a.  $\psi = \sin x$
  - b.  $\psi = \tan x$
  - c.  $\psi = \operatorname{cosec} x$
  - d.  $\psi = x$
- 2. The ratio  $N_n=2: N_n=1$  [ $N_n$  is the number of nodes in a state with quantum number n for a particle in a 1-D box] is

a. 0 <b>.</b>	c. 2
b. 1	d. zero

- 3. The energy of a particle in a 2-D box of size 'A' is 9h<sup>2</sup>/4ma<sup>2</sup>. The degree of degeneracy is
  a. 1
  c. 3
  - b. 2 d. 4
- 4. Which of the following is NOT a physical requirement for an acceptable wave function.a. Symmetricc. Square integrable
  - b. Single valued d. Continuous in the given region
- 5. The speed of the de brogile wave of a particle of mass 'm' and momentum mv is
  - a.  $\frac{c^2}{v}$ b.  $\frac{c}{v^2}$ [C= velocity light]
- 6. The energy required to excite a particle of mass 'm' confined in a length 'l' to the first excited state is

a.	$h^2/2ml^2$	c. $h^2/4ml^2$
b.	3h <sup>2</sup> /8ml <sup>2</sup>	d. l <sup>2</sup> /2mh <sup>2</sup>

- 7. An orbital is a/an
  - a. operator

b. one electron wave function

c. circular tract d. observable property

8.	Conjugate of an op	perator $\hat{A} = \hbar \frac{d}{d}$	$\frac{d}{dx}$ is $\hbar \frac{d}{dx}$ . Then conj	ugate of the operat	or
	$\hat{o} = i\hbar \frac{d}{dx}$				
	a sta		a B d		
	$\int_{-\frac{d}{dx}}^{d} dx$		d. $-\hbar \frac{d}{dx}$		
9.	dx Two functions are	orthogonal if th	dx eir inner product is		
	a. n 2		c1		
	b. Zero		d. 1		
10.	The eigenvalue of	$\widehat{P}_{21}$ is			
	a. Zero		c. +1		
	b. ± 1		d1		

**11.** The operator of kinetic energy in one dimension is

a. $i\hbar \frac{\partial}{\partial x}$	ð			$\hbar^2$	$\partial^2$
	c. $-\frac{1}{2m}$	-2m	$\partial x^2$		

- b.  $-i\hbar \frac{\partial}{\partial x}$  d.  $\frac{\hbar^2}{2m} \frac{\partial^2}{\partial x^2}$
- 12. The commutation relation between position and momentum operator is **a.**  $\left[q_{ij}, p_k\right] = 0$ 
  - $\begin{array}{l} \text{c.} & \left[ q_{ij}, p_k \right] = i \hbar \delta_{jk} \\ \text{d.} & \left[ q_{ij}, p_k \right] = -i \hbar \delta_{jk} \end{array}$

b.  $[q_{ij}, p_k] = 1$ 

- 13. If  $\widehat{P_{y}} = -i\hbar\frac{\partial}{\partial y}$  then  $\widehat{P_{y}^{3}}$  is a.  $-i\hbar^{3}\frac{\partial^{3}}{\partial y^{3}}$ 
  - b.  $-i\hbar \frac{\partial^3}{\partial y^3}$ d.  $i\hbar \frac{\partial^3}{\partial y^3}$

c.  $i\hbar^3 \frac{\partial^3}{\partial y^3}$ 

2017/12

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<ul><li>14. The exchange energy between two ele</li><li>a. Same spin</li><li>b. Opposite spin</li></ul>	ectrons is related when they have c. Same spin in degenerate orbital d. Opposite spin in non degenerate orbital.	[PART (A) : OBJECTIVE]     Serial no. of the main Answer sheet       Duration : 20 Minutes     Serial no. of the main Answer sheet
<b>15.</b> 2 s orbital of hydrogen atom has a no	de at $2a_0$ , because $\psi_{2s}$ is proportional to	
a. $(2 + \frac{r}{a_0})$	c. $(1 + \frac{r}{2a_0})$	Course :
b. $(2 - \frac{r}{2a_0})$	d. $(2 - r/a_0)$	Semester : Roll No :
16. Two operators $\widehat{A}$ as $\widehat{A}(c\psi) = c(\widehat{A}\psi) and \widehat{B}\psi = \psi *$	nd $\widehat{B}$ satisfy the equations (c is a constant).	Enrollment No : Course code :
Select the correct statement from	n the following:	
a. $\widehat{A}$ and $\widehat{B}$ are Hermitian	c. $\widehat{B}$ is linear but $\widehat{A}$ is not	Course little :
b. $\widehat{A}$ and $\widehat{B}$ are commute	d. $\widehat{A}$ is linear but $\widehat{B}$ is not	Session : 2017-18 Date :
<ul><li>17. The lowest energy is zero for a</li><li>a. Hydrogen atom</li><li>b. Particle in 1-D box</li><li>c. Simple harmonic oscillator</li><li>d. Particle in a ring</li></ul>		<ul> <li>Instructions / Guidelines</li> <li>The paper contains twenty (20) / ten (10) questions.</li> </ul>
<b>18.</b> Which of the following obey Fermi-D	irac statistics?	> Students shall tick ( $\checkmark$ ) the correct answer.
a. Pion $(\pi^+)$ b. Kaon $(k^+)$	c. η meson d. Muon (μ <sup>-</sup> )	<ul> <li>No marks shall be given for overwrite / erasing.</li> <li>State of the state of</li></ul>
<b>19.</b> The expectation value of the observal a. $< 0 >= \int_{-1}^{1} \psi * \psi d \Upsilon$	ble 0 is given by c. $< 0 >= \int_{-\infty}^{\infty} \psi \hat{0} \psi d \mathcal{F}$	Students have to submit the Objective Part (Part-A) to the invigilator just after completion of the allotted time from the starting of examination.
b. $< 0 >= \int_{*}^{\infty} \psi * \hat{0} \psi  dY$ [ $\hat{O}$ = operator of the observable	$d. < 0 >= \int_{-\infty}^{\infty} \psi  \hat{0}  \psi  d  Y$	Full Marks Marks Obtained
<b>20.</b> If the operator $\widehat{A}$ commutes with the	operator $\widehat{B}$ then	20

- a.  $\widehat{A}$  and  $\widehat{B}$  are equal
- b.  $\widehat{A}$  and  $\widehat{B}$  are skew- hermitian
- <sup>c.</sup>  $\widehat{A}$  and  $\widehat{B}$  have common set of eigen values <sup>d.</sup>  $\widehat{A}$  and  $\widehat{B}$  have the common set of eigenfunctions.

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Scrutinizer's Signature

Examiner's Signature

Invigilator's Signature