

B.Sc. CHEMISTRY FOURTH SEMESTER ORGANIC CHEMISTRY-III

BSC-402 [REPEAT] [USE OMR FOR OBJECTIVE PART]

Duration: 3 hrs.

Time: 30 min.

Objective)

Marks: 20

Full Marks: 70

2023/06

SET

Choose the correct answer from the following:

1X20 = 20

- Which of the following methods is used to prepare nitriles?
 - a. Hydration of amines
- b. Dehydration of acids
- c. Reduction of acids
- d. Dehydration of amides
- When acetamide is allowed to react with $Br_2/NaOH$ it results in the formation of
 - a. Ethylamine

b. Urea

c. Methylamine

- d. Acetyl bromide
- Acetonitrile when is subjected to reduction with LiAlH₄ it gives
 - a. Methylamine

b. Dimethylamine

c. Ethylamine

- d. Trimethylamine
- Nitromethane on being reduced with Sn/HCl gives
 - a. Methyl amine

b. Ethyl amine

c. n-Propylamine

- d. Isopropyl amine
- 5. The product P in the following reaction is

- a. Chlorobenzene
- c. Phenol

- b. Aniline
- d. Nitrobenzene
- The presence of -N-CH₃ group in an alkaloid can be established by
 - a. Herzig-Meyer reaction
- b. Hofmann reaction

- c. Emde's reaction
- d. Sandmeyer reaction
- The alkaloid used as sedatives is
 - a. Hygrine

b. Morphine

c. Reserpine

- d. Nicotine
- An alkaloid is known to have used to relieve pain. The name of the alkaloid is
 - a. Nicotine

b. Morphine

c. Reserpine

d. Hygrine

9.	Nicotine when is treated with $ZnCl_2$, zincichloride is produced. This product on being treated with lime gives		
	Pyridine, pyrrole and methylamine a.	b.	Pyrrolidine, piperdine, and methylamine
	Pyridine, pyrrole and ethylamine	d.	Pyrrolidine, piperdine, and ethylamine
10.	Nicotine belongs to the class of a. Pyrrolidine alkaloid c. Pyridine-pyrrolidine alkaloid		Pyridine alkaloid None of these

11. When Benzene and Phthalic anhydride reacts, it produces b. Phenanthrene a. Anthracene

c. Naphthalene

d. Pyridine 12. Naphthalene when treated with conc. HNO₃ and conc. H₂SO₄ it produces majorly

a. 1-nitronaphthalene c. 3-nitronaphthalene

b. 2-nitronaphthalene d. 1-aminonaphthalene

13. β -naphthol when treated with NH₃, NaHSO₃ gives

a. a-naphthylamine

b. β-naphthylamine

c. γ-naphthylamine

d. None of these

14. When Benzene reacts with anhydride of succinic acid it gives

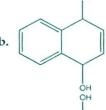
a. Naphthalenec. Pyrrole

b. Pyridine

d. Anthracene

15. CrO₃, AcOH

The product A is



[2]

USTM/COE/R-0

The product P is a. Phenanthrene

c. Naphthalene

17. THF correspond to the structure

18. Indole is

a. Non-aromatic

c. Aromatic

19. Pyridine is

a. more basic than pyrrole

c. equally basic with pyrrole

20. Hantzsch synthesis is used to preparea. Naphthalenec. Pyridine

b. Anthracene

d. None of these

b. Anti-aromatic

d. None of theses

b. less basic than pyrrole

d. acidic in nature

b. Benzened. Formaldehyde

(<u>Descriptive</u>)

Time: 2 hrs. 30 mins.

Marks: 50

[Answer question no.1 & any four (4) from the rest]

1. a. Describe Haworth synthesis Naphthalene.

3+2+2+3=10

b. Write down the products A and B

c. Explain Hofmann elimination. What are the products formed when Hofmann elimination is carried out with piperidine?

2. a. Write down the products A, B, C and D

4+3+3=10

3+4+3

=10

b. Convert 2-naphthol to Naphthalene-2-carboxylic acid

c. How will you prepare Naphthalene-1-carboxylic acid from (i) naphthalene and (ii) 1-Naphthylamine.

3. a. What will happen when anthracene reacts with maleic anhydride?

 b. Show the synthesis of anthracene using benzene and Phthalic anhydride.

c. Mention all the compounds A, B and C

$$+ \longrightarrow A \xrightarrow{AcOH, CrO_3} B \xrightarrow{Zn/HCl} C$$

[4]

USTM/COE/R-01

- 4. a. Write down the structure of pyrrole, thiophene, imidazole, oxazole and pyrazole.
 - 5+2+3 =10
 - b. Furan is aromatic in nature but tetrahydrofuran is not aromatic-Why?
 - c. Describe paal-knorr synthesis of thiophene.
- a. Write down the structure of Indole. Show the preparation 5. of indole using Fischer indole synthesis.
 - 1+4+2+1 +2 = 10
 - b. Write one method of preparation of nitroalkane. What will be the product formed when nitromethane is treated with Cl2 and NaOH? Identify the products A and B in the following sequence of reactions

$$CH_3CH_2CH_2Br + AgCN \longrightarrow A \xrightarrow{H^+/H_2O} Hydrolysis$$

- a. How would you distinguish between primary, secondary, and tertiary amines with nitrous acid? Explain with chemical reactions.
 - 3+3+4 =10
 - b. Justify the following order of the basic strength of amines based on inductive effect, hydration effect and steric effect (CH₃)₂NH > CH₃NH₂ > CH₃)₃ N
 - Complete the following reactions and identify the products formed.

(i)
$$CH_3CH_2CN + CH_3MgBr$$
 (ii) H_2O Product

(ii)
$$MgBr + CH_3CH_2CN \xrightarrow{\text{(i) Ether}} Product$$

a. Complete the following reaction and name of the product

2+3+5 =10

5+5=10

- b. What are alkaloids? Discuss the general features of alkaloids?
- Write the structure of hygrine. To which class does it belong? How did Hess synthesize hygrine starting from pyrrylmagnesium bromide? Explain.
- How will you show that, hygrinic acid obtained from hygrine is N-methyl pyrrolidine-2-carboxylic acid? Write the Willstatter's synthesis of hygrinic acid.
- Write the structure of nicotine. How will you suggest that

 (i) nicotine contains one -N-CH₃ group and (ii) the side
 chain of nicotine is saturated?

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