B.Sc. CHEMISTRY THIRD SEMESTER ORGANIC CHEMISTRY II BSC - 302 [SPECIAL REPEAT] [USE OMR FOR OBJECTIVE PART]

SET

Full Marks: 70

Duration: 3 hrs.

Objective

Time: 30 min.

Marks: 20

Choose the correct answer from the following:

1X20=20

- Benzene diazonium chloride upon reaction with aqueous solution produce
 - a. benzene

b. Bromobenzene

c. catechol

- d. Phenol
- Reimer-Tieman reaction proceed via the formation of
 - a. free radical

b. carbocation

c. carbanion

- 3. The correct order of the acidic strength of the following:

a. I>II>III

b. 111>1>11

c. III>II>I

- d. |>|||>||
- Hoffmann Bromamide reaction convert
 - a. primary amide to a primary amine
 - c. primary amide to an aldehyde
- b. primary amide to a primary alcohol
- d. primary alcohol to a primary amine
- 5. The substrate which can undergo Cannizzaro reaction is
 - a. Enolizable aldehyde
- b. Non-enolizable aldehyde

c. Enolizable ketone

- d. Non-enolizable ketone
- 6. The intermediate compound of Aldol reaction is
 - a. a -Hydroxy ester

b. a-Hydroxy carbonyl

c. β-Hydroxy ester

- d. β-Hydroxy carbonyl
- 7. The compound which can be used as Michael acceptor is
 - a. α-β unsaturated carbonyl
- b. β-y unsaturated carbonyl
- c. γ-δ unsaturated carbonyl
- d. All of these
- 8. The alcohol which does not undergo oxidation with PDC is
 - a. isopropanol

b. tertiary butanol

c. benzyl alcohol

d. Ethanol

- 9. What is the main product obtained by the oxidation of thioethers with KMnO₄
 - a. Sulphone

b. Sulphide

c. Thioacetals

- d. Sulphoxide
- 10. Which is the main product of the following reaction?

$$R-MgX + S \longrightarrow R-S-MgX \xrightarrow{R'X} (?) + MgX_2$$

a. Thiols

b. Thioethers

c. alkane

- d. Halo acids
- 11. For preparation of any p-alcohol using Grignard reagent (GR), need
 - a. GR + HCHO

b. GR + RCHO

c. GR + Ketone

- d. GR + H₂O
- 12. An organic compound A reacts with sodium metal and forms B. On heating with conc. H₂SO₄, A gives diethyl ether. A and B are respectively
 - a. C₂H₅OH and C₂H₅ONa
- b. C2H5OH and CH3ONa
- c. C₃H₇OH and C₃H₇ONa
- d. CH₃OH and C₂H₅ONa
- 13. The major product in the given reaction

- 14. Which is a not a correct statement for an SN² reaction?
 - a. Proceed with inversion of configuration b. strong nucleophile will favour the reaction.
 - reaction.
 - c. weak nucleophile will favour the d. polar aprotic solvent favour SN2 mechanism.
- 15. A and B in the following sequence of reactions are respectively

- a. ROCI and ROMg
- b. RCI and RMgCI

- c. RCl and RMgOCl
- d. ROCI and RCI
- 16. Strength of the following nucleophile will be of the order:
 - a. CH₃S-> CH₃O-> t-BuO-
- b. CH₃O-> CH₃S-> t-BuO-
- c. t-BuO-> CH3S-> CH3O-
- d. CH₃S-> t-BuO->CH₃O-

17. The example of a trihydric alcohol is

a. glyceraldehyde

b. glycerol

c. ethylene glycol

d. ethanol

18. Acetone when reacts with methyl magnesium bromide (CH₃MgBr) produce

a. 1° alcohol c. 3° alcohol

b. 2° alcohol d. no reaction

19. The main reagent for Hydroboration - Oxidation reaction of alkene is b. H₂/Pd

a. B₂H₆ c. CrO₃-H₂SO₄

d. Hg(OAc)2

20. When cyclohexene is treated with mCPBA, the product is a/an

a. alcohol c. aldehyde b. epoxide d. ketone

Descriptive

Time: 2 hrs. 30 mins.

Marks:50

3

[Answer question no.1 & any four (4) from the rest]

1. a. What are the products of the given reactions? explain

b. Why the following reaction gives such results?

c. i. Discuss the preparation of salicylaldehyde starting from phenol *via* Riemer-Tiemann reaction?

ii. What are the products of the following reaction and which one is the major product?

2. a. Write down the product / products formed in the following reaction with mechanism.

b. Write down the products A, B and C in the following sequence of reactions

c. Sketch out a strategy for synthesis of the following compounds using CH_3CH_2Br

- 3. a. Discuss the stereochemical changes that may be observed when an optically active substrate undergoes nucleophilic substitution following SN¹ mechanism.
 - b. "A strong nucleophile favour SN² mechanism and weak nucleophile favours an SN¹ mechanism." Justify the statement.
 - c. Predict the products in the following reactions:

- 4. a. How will you prepare ethanol from acetaldehyde. Why is the boiling point of ethanol greater than acetaldehyde-explain schematically?
 - b. Write down the structure of picric acid, 2-nitrophenol and 2-
 - ethylphenol and comment which one is more acidic in nature.

 c. Describe Pinacol-pinacolone rearrangement.
 - d. How can you prepare epoxide from halohydrins? Explain with a suitable example

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5. a. Write the reagents (A & B) of the following reactions.

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2

OH SOCI₂ A
$$\frac{H_2}{Pd/BaSO_4}$$
 B Quiniline

b. Write a short note on Cannizzaro reaction.

c. What product will be formed if CO₂ is treated with 2 equivalents of PhLi in anhydrous THF followed by acidic treatment?

6. a. Write down the products of the following reactions

b. Show the mechanistic pathway of Curtius Rearrangement.

c. Write down the structure of Maleic acid and Fumaric acid. Which one can form corresponding anhydride easily upon heating via intramolecular dehydration? Draw the structure of that anhydride.

d. How will you prepare Phthalic anhydride from Phthalic acid?

7. a. Why 4-nitrophenol has greater boiling point than 2-nitrophenol - draw the structures in support of your answer?

b. Describe Kolbe-Schmitt reaction and show the mechanism for the formation of salicyclic acid from phenol.

c. Write down the products

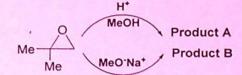
d. Write down the short note on Fries Rearrangement with detailed mechanism? What is the role of AlCl₃ in the reaction?

HSTM/COF/R-01

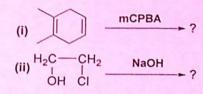
8. a. Identify 'A' & 'B'? Show most appropriate mechanistic pathway for the formation of A & B

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3+2=5



b. What will be the product of the following reaction



c. Identify 'A' & 'B' of the following reactions.

(i) SH SH & HCI Raney Ni / H₂ B

(ii) B $\stackrel{\text{H}_2\text{O}_2}{\longrightarrow}$ Me $\stackrel{\text{KMnO}_4}{\longrightarrow}$ A