B. Sc. CHEMISTRY FOURTH SEMESTER ORGANIC CHEMISTRY III

BSC - 402

(Use Separate Answer Scripts for Objective & Descriptive)

Duration: 3 hrs.

Time: 20 min.

Full Marks: 70

(PART-A: Objective)

Marks: 20

Choose the correct answer from the following:

1X20 = 20

- 1. Naphthalene is a
 - a. Isolated benzenoid system
 - c. Fused benzenoid system
- 2. Pyridine is a
 - a. weak base
 - c. neutral molecule
- 3. The nature of Indole is
 - a. Non-aromatic
 - c. Anti-aromatic

- b. Fused non-benzenoid system
- d. Isolated benzenoid system
- b. weak acid
- d. Strong base
- b. Aromatic
- d. Aliphatic
- 4. Common name of the following structure is

- a. Pyrrole
- c. pyrazole

- b. Imidazole
- d. oxazole
- 5. Molecular formula of isoprene unit is
 - a. C₅H₈
 - c. C5H5

- b. C₅H₁₀
- d. C₅H₁₂
- 6. The correct nomenclature of the following compound is

- a. Naphthalene-2,7- disulfonic acid
- c. Naphthalene-3,7- disulfonic acid
- 7. The structure of Naphthalene has
 - a. 4α and 2β position
 - c. 2a and 2ß position

- b. Naphthalene-2,8- disulfonic acid
- d. Naphthalene-1,7- disulfonic acid
- b. 4α and 4β position
- d. 2α and 4β position

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8.	Anthracene is a a. Linear non-aromatic system c. Non-linear benzenoid fused system		Linear benzenoid fused system Non-linear non-benzenoid fused system
9.	The structure of THF is a.	b.	S
	c. N	d.	Se
10.	Hantzsch synthesis is related to a. Anthracene c. Benzene		Phenanthrene Pyridine
11.	 Alkaloids are a. natural products, founds in animals, that heteroatoms c. natural products, founds in plants, that atoms d. alcohol-like compounds 	t co	ntains one or more nitrogen
12.	Quinine is obtained from the bark of which a. Cinchona c. Banyan	b.	e? Redwood Eucalyptus
13.	Nicotine causes which of these changes in ta. lowers body temperature c. increases blood pressure	b.	ody? increases heart rate increases heart rate and blood pressure
14.	The alkaloid used as sedative is a. Cocaine c. Morphine		Hygrine Atropine
15.	What happens when nicotine comes into co a. tissues swell c. nerve centers shut down	b.	ct with the brain? dopamine is released adrenaline increases
16.	The alkaloid Quinine is used in the treatme a. Malaria c. Parkinson disease	b.	f High blood pressure Alzheimer
17.	Nitriles can be prepared by a. the hydration of amines c. the reduction of acids		the dehydration of acids the dehydration of amides
	[2]		USTM/COE/R-01

18. Acetamide reacts with Br₂/NaOH to give

a. Methylaminec. Ethylamine

b. Urea

d. Acetyl bromide

19. Acetonitrile undergoes reduction with LiAlH4 to form

a. Methylaminec. Ethylamine

b. Dimethylamine

d. Trimethylamine

20. Which type of amines produces N_2 when treated with HNO_2 ?

a. Quaternaryc. Secondary

b. Tertiaryd. Primary

PART-B: Descriptive

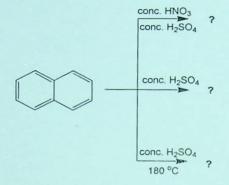
Time: 2 hrs. 40 min.

Marks: 50

[Answer question no.1 & any four (4) from the rest]

1. a. Write down the products:

5+5=10



- b. What is Gabriel phthalimide synthesis? How will you synthesize aliphatic primary amine by this method? Explain why aromatic primary amine cannot be prepared by this method?
- 2. a. What is Citral? Draw the structure of *cis* and *trans*-form of Citral. Describe its synthesis.
 - **b.** Draw the structure of α-terpineol. Describe the synthesis procedure and mention its uses
- 3. a. Explain Haworth synthesis of naphthalene

6+4=10

5+5=10

b. Write down A, B and C

[4]

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4. a. How the name 'indole' given? Describe Fischer indole synthesis 5+5=10 with detailed mechanism. b. Write down short note on Knorr-quinoline synthesis and its mechanism a. Draw the structures of pyrazole and oxazole. Describe the Knorr 5+5=10 pyrrole synthesis with mechanism. b. What is benzene diazonium salts? How are they produced? Starting from benzene diazonium chloride how will you synthesize (i) chlorobenzene and (ii) phenol. Give chemical 5+5=10 6. a. What are alkaloids? Discuss the general features of alkaloids. b. How are alkaloids isolated from plants? Describe their general properties. a. What is Mannich reaction? Write the mechanism of the reaction. 5+3+2 =10Can you use an aldehyde other than formaldehyde? b. Write a note on Hofmann's elimination reaction. c. How does nitrous acid react with primary amine? Give the chemical equation. 5+5=10 8. a. Discuss the effect of substituent and solvent on the basicity of amines. Give reason why order of the basic strength in case of methyl substituted amines in aqueous solution is as follows: $(CH_3)_2NH > CH_3NH_2 > CH_3)_3 N > NH_3.$ b. Write the carbylamine reaction. Explain why this reaction can be used as a test for only primary amine but not for secondary and

tertiary amine?