## M.Sc. CHEMISTRY THIRD SEMESTER ORGANIC CHEMISTRY-III MSC-301

Duration: 3 Hrs.

Marks: 70

PART: A (OBJECTIVE) = 20 PART: B (DESCRIPTIVE) = 50

[ PART-B : Descriptive ]

Duration: 2 Hrs. 40 Mins.

Marks: 50

#### [Answer question no. One (1) & any four (4) from the rest]

1. Answer the following questions:

 $(2 \times 5 = 10)$ 

- (a) Explain, what do you mean by inversion of sucrose?
- (b) Discuss the role of Chlorophyll in photosynthesis.
- (c) Write down the product of the following reaction with mechanism.

(d) Find the iso-electric point of the following amino acids.

(i) 
$$HOOC \longrightarrow C \longrightarrow CH_2COOH$$
 (ii)  $H_3N - C - C - C - C - CH \longrightarrow COOH$   $hH_2 \longrightarrow pka = 8.95$   $hH_2 \longrightarrow pka = 8.95$   $hH_3 \longrightarrow pka = 8.95$   $hH_2 \longrightarrow pka = 2.18$ 

(e) Find out the order of following sigma tropic rearrangements:

2. (a) Write down the structure of the products A, B, C and D.

(5+3+2=10)

Maltose 
$$\xrightarrow{\text{Br}_2 \text{ water}}$$
 A  $\xrightarrow{\text{Me}_2 \text{SO}_4 / \text{NaOH}}$  B  $\xrightarrow{\text{H}_2 \text{O} / \text{H}}$  C + D

What information you can get from this result about the structure of maltose.

(b) Complete the following reactions:

(i) Glucose	MeOH / HCI		?
(ii) Glucose	Ac <sub>2</sub> O		
	NaOAc		•

- (c) Write down the products of oxidation reaction of sucrose with periodic acid.
- 3. (a) What is Ninhydrin reagent and what is its use? Explain with suitable (3+3+4=10) example.
  - (b) What do you meant by sequencing of amino acid? What is the method used for N-terminal sequence determination of amino acids in a peptide chain?
  - (c) What is secondary structure of protein and how it is classified? Explain any one of the stable arrangements of secondary structure of protein.
- 4. (a) Discuss the role of hemoglobin as oxygen carrier and explain its (4+(3+2)+1=10) oxygen binding mechanism.
  - (b) Explain various methods of preparation of dipyrrylmethenes and hence suggest at least one useful method to synthesize porphyrin from dipyrrylmethenes.
  - (c) What are phthalocyanines?
- 5. (a) Write down the products (A & B) with explanation.

(3+3+(2+2)=10)

(b) Write down the products (A & B) with explanation.

(c) Write down the products (A & B) for the following reactions.

6. (a) Predict the products of the following reactions:

(2+2+2+4=10)

(i) 
$$H_3C$$
  $CH_3$   $\Delta$  (ii)  $CH_3$   $CH_3$   $CH_3$ 

- (b) What is 1, 3-dipole? Give example of it.
- (c) Why [2+2] cycloaddition reaction does not take place under thermal condition? Explain.
- (d) Match the following:

(i) 
$$\parallel$$
 +  $\parallel$  — (a) Chelatropic reaction

(ii)  $\parallel$  +  $SO_2$  —  $\parallel$   $SO_2$  (b) Cycloaddition reaction

(iii)  $\parallel$  R — R — (c) Electrocyclic reaction

(iv)  $\parallel$  — (d) Sigmatropic reaction

- 7. (a)  $\alpha$ -maltose show specific rotation of (+) 168° and  $\beta$ -maltose (+) 118°. (2+3+3+2=10) Both these anomers of maltose exhibit mutarotation and specific rotation shown when the equilibrium is reached is (+) 136°. Calculate the percentage of  $\alpha$ -maltose and  $\beta$ -maltose at equilibrium.
  - (b) Write a note on structural differences between starch and cellulose.
  - (c) Show by the correlation diagram that the thermal Diels Alder reaction is an allowed cycloaddition reaction.
  - (d) What is Ene reaction? Explain with suitable example.

(a) Write down the reagent and the products for following reactions:

(b) Explain the linkage of peptide. How the tripeptide Gly-Ala-Gly can (2+3=5)be synthesized? Explain with suitable reactions.

1×20=20

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## [PART-A: Objective]

#### Choose the correct answer from the following:

1. Structure of  $\beta$ -maltose is:

- 2. Which of the following statement is not correct?
  - a. Starch is a mixture of two polymers.
  - b. Cellulose is a polymer of glucose linked by  $\beta$ -1-4 glycosidic linkage.
  - c. Chitin is a polymer of N-acetyl glucosamine.
  - d. Glycogen is an animal sugar structurally similar to amylose.
- 3. Which of the following will not form an osazone?
  - a. Maltose
- b. Fructose
- c. Sucrose
- d. Lactose

- 4. Glucose and mannose are:
  - a. enantiomers
- b. epimers
- c. anomers
- d. none of these

- 5. Which is a correct statement for lactose?
  - a. A disachharide of galactose and glucose with  $\alpha$ -1-4 glycosidic linkage.
  - b. A disachharide of glucose and galactose with α-1-4 glycosidic linkage.
  - c. A disachharide of galactose and glucose with  $\beta$ -1-4 glycosidic linkage.
  - d. A disachharide of glucose and galactose with  $\beta$ -1-4 glycosidic linkage.
- 6. Electrophilic substitution reaction of N-alkylated-pyridine occurs regio-specifically at:
  - a. α-carbon
- **b.** β-carbon
- c. y-carbon
- d. non-regio-specific
- 7. Quinoline reacts with benzoylchloride to give:

- 8. Chichibabin reaction of pyridine gives the product of:
  - a. C-alkylation

b. C-hydroxylation

- c. C-amination
- d. C-nitration
- 9. Regio-selective nitration of indole at  $\beta$ -carbon can be obtained by using:
  - a. Conc. HNO<sub>3</sub>
- b. mixture of Conc. HNO<sub>3</sub> and Conc. H<sub>2</sub>SO<sub>4</sub>
- c. Conc. HNO<sub>3</sub> and Ac<sub>2</sub>O
- d. PhCOONO<sub>2</sub>
- 10. When the amino acid alanine is added to a solution with a pH of 7.3, it becomes:
  - a. A cation

b. Non polar

c. An isotope

- d. A zwitter ion
- 11. The reagent used for deprotection of the following protecting group:

a. Hydrofluoric acid

- b. Piperidine
- c. H<sub>2</sub> and a Pd charcoal catalyst
- d. Trifluoroacetic acid

12. The  $\alpha$ -helix rises per turn a distance of:

a. 0.54 nm

b. 1.5 nm

c. 3.0 nm

d. 1.83 nm

13. Which of the following is a set of hydrophobic amino acids?

a. Arginine-Lysine-Metheonine

b. Aspartic acid-Valine- Isoleucine

c. Histidine-Proline-Glutamine

d. Valine-Metheonine-Isoleucine

14. The Claisen rearrangement is:

a. [1,5] sigmatropic rearrangement

b. [3,3] sigmatropic rearrangement

**c.** [1,3] sigmatropic rearrangement

d. None of the these

15. The product of the following reaction is:

- d. None of these
- 16. Sommelet Hauser rearrangement takes place through:
  - a. [3,3] sigmatropic rearrangement
- b. [2,3] sigmatropic rearrangement
- c. [1,3] sigmatropic rearrangement
- d. None of these
- 17. The following conversion takes place through:

- a. [4+2] Cycloaddition reaction
- b. [2+2] Cycloaddition reaction
- c. [8+2] Cycloaddition reaction
- d. None of these

18. Which is not the correct statement as regards to binding of hemoglobin with ligands?

- a. Oxyhemolgobin is extremely stable and the coordination is irreversible.
- b. Oxyhemolgobin is stable and the coordination is reversible.
- c. Coordination is irreversible in case of cyanide ion.
- d. Cyanide ion has much higher affinity to bind with Fe than that of Oxygen.

19. The correct molecular formula of Porphin is:

- a. C<sub>20</sub>H<sub>14</sub>N<sub>4</sub>
- b. C<sub>20</sub>H<sub>12</sub>N<sub>4</sub>
- c. C<sub>18</sub>H<sub>20</sub>N<sub>4</sub>
- d. C<sub>22</sub>H<sub>20</sub>N<sub>4</sub>

**20.** The four protein chains of hemoglobin comprises of two  $\alpha$  chains and two  $\beta$  chains. The number of residues of each of the two chains  $\alpha$  and  $\beta$  respectively are:

- a. 141 and 146
- **b.** 142 and 145
- c. 140 and 147
- d. 150 and 151

# **UNIVERSITY OF SCIENCE & TECHNOLOGY, MEGHALAYA**



### [PART (A) : OBJECTIVE]

**Duration: 20 Minutes** 

Seri	al no. of the
main	Answer sheet

Course:		
Semester:	Roll No :	concernation contents
Enrollment No:	Course code :	
Course Title :		
	7-18 Date :	
<ul> <li>Students shall tick (</li> <li>No marks shall be g</li> <li>Students have to sub</li> </ul>	Instructions / Guidelines  twenty (20) / ten (10) questions.  /) the correct answer.  given for overwrite / erasing.  bomit the Objective Part (Part-A) to the invigil  llotted time from the starting of examination.  Full Marks Marks Obtained  20	ator just after
Scrutinizer's Signature	Examiner's Signature	Invigilator's Signature